PRECIPITATION OF POORLY SOLUBLE DRUG DURING DISSOLUTION

DETERMINED BY MRI AND ATR-FTIR IMAGING

Kateřina Punčochová a, b, Josef Beránek b, Andrew Ewing c, Sergei G. Kazarian c *, František Štěpánek a *

- ^a Department of Chemical Engineering, Institute of Chemical Technology Prague, Czech Republic
- ^b ZENTIVA k.s., U Kabelovny 130, Prague 10, Czech Republic
- ^c Department of Chemical Engineering, Imperial College London, Great Britain
- *Frantisek.Stepanek@vscht.cz
- *s.kazarian@imperial.ac.uk



Aim of the study

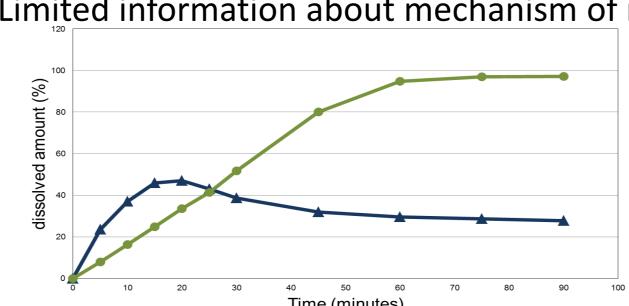
- Enhancement of bioavailability of poorly soluble drugs
- **Observation of drugs** dissolution by ATR-FTIR Imaging and MRI
- Mechanism of dissolution
- **Precipitation**

Dissolution test

- Concentration of API as a function of time
- Rate of drug release
- Limited information about mechanism of release

Attenuated total reflection (ATR) - FTIR Imaging

Bruker Equinox mid-IR imaging system in ATR mode ^{1,2}



Introduction

Imaging methods

- What happened during dissolution?
- Selection of candidate formulations
- Effect of polymers, additives

Precipitation of amorphous form to crystalline form

- local supersaturation of diffusing drug leads to precipitation of drug
- decreasing of bioavailability

Dissolution of tablet Flow Dissolving layers Concentration profile of polymer

Imperial College

London

Materials

534.4 g/mol

Aprepitant (Drug)

- molecular weight
- logP
- 3.5, 9.6 pKa
- solubility in water (20 °C) 0.02 mg/ml

Polyvinyl pyrrolidone (PVP)

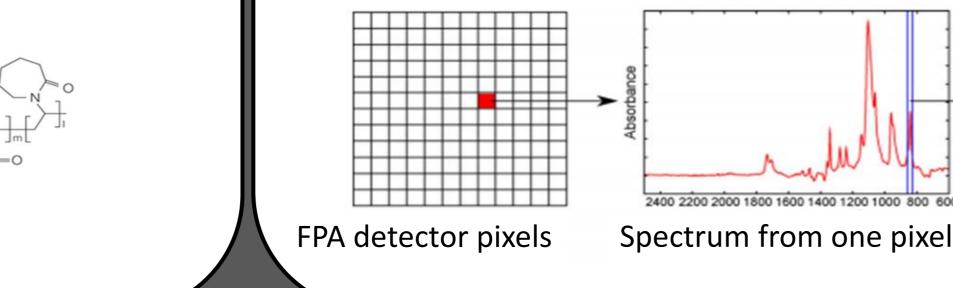
- hydrophilic polymer
- soluble in water

Solid dispersion preparation

- drug dispersed in polymer matrix
- preparation by spray drying
- ratio 1:3 (drug:polymer)
- amorphous form of drug

Soluplus

- amphiphilic polymer
- colloidal micelles in water



Flow of water

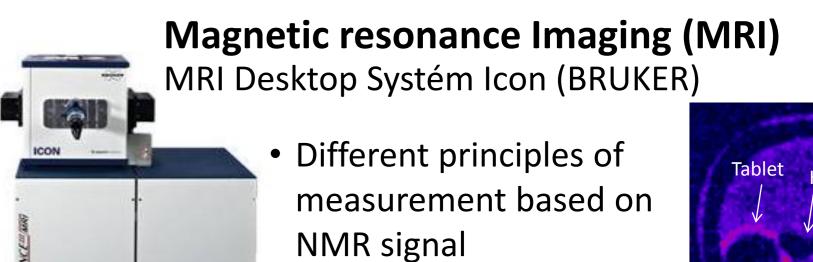
Imaging methods

Water

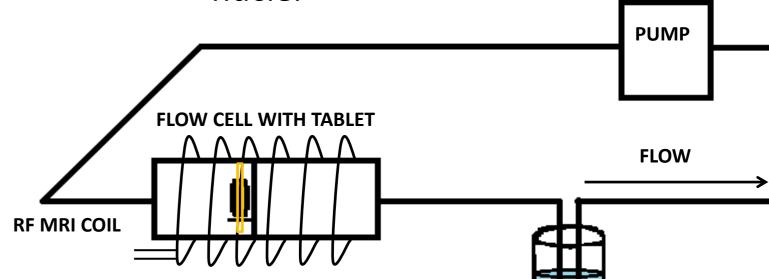
Tablet

Integrated absorbance

plotted for all pixels



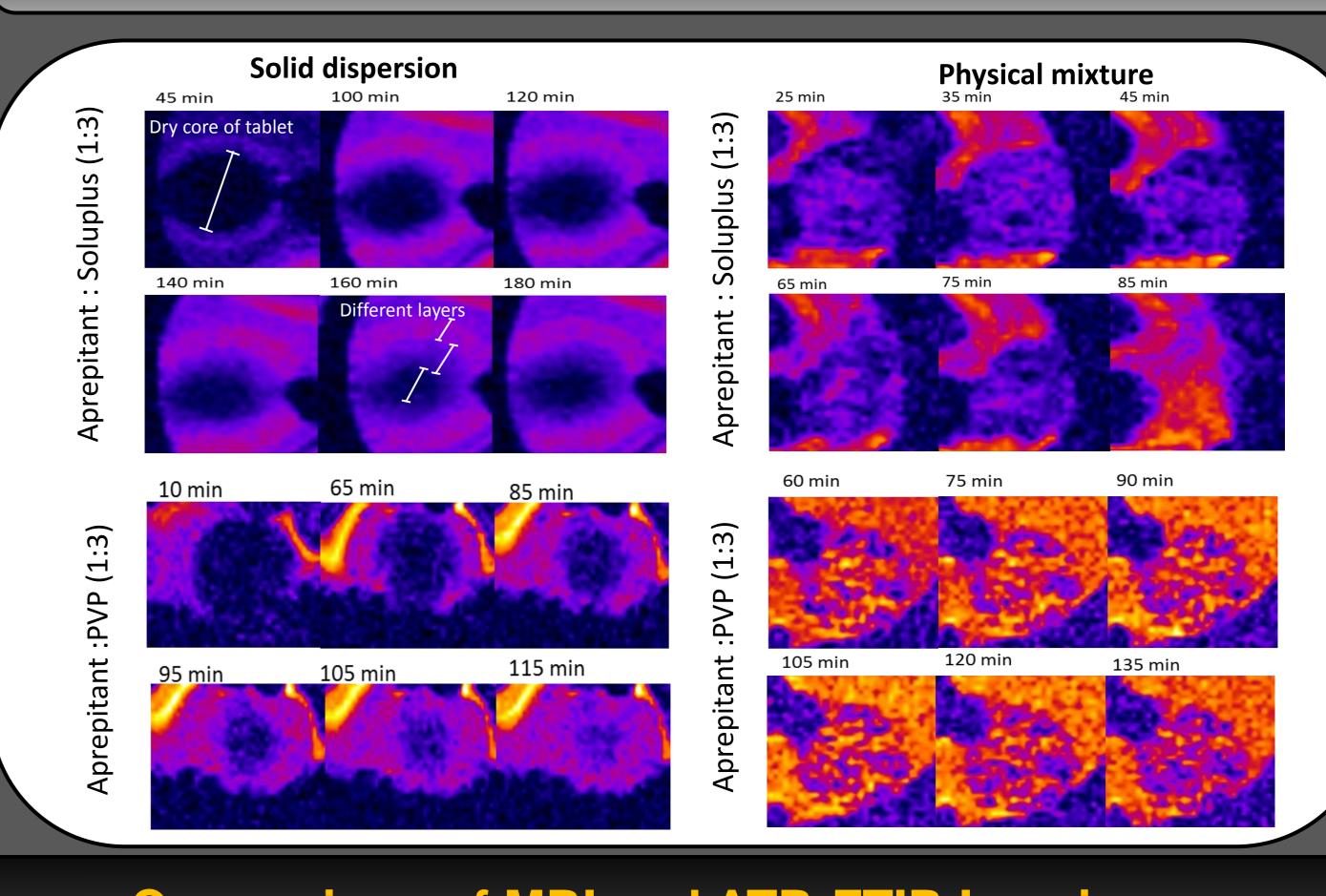
• Concentration of ¹H nuclei



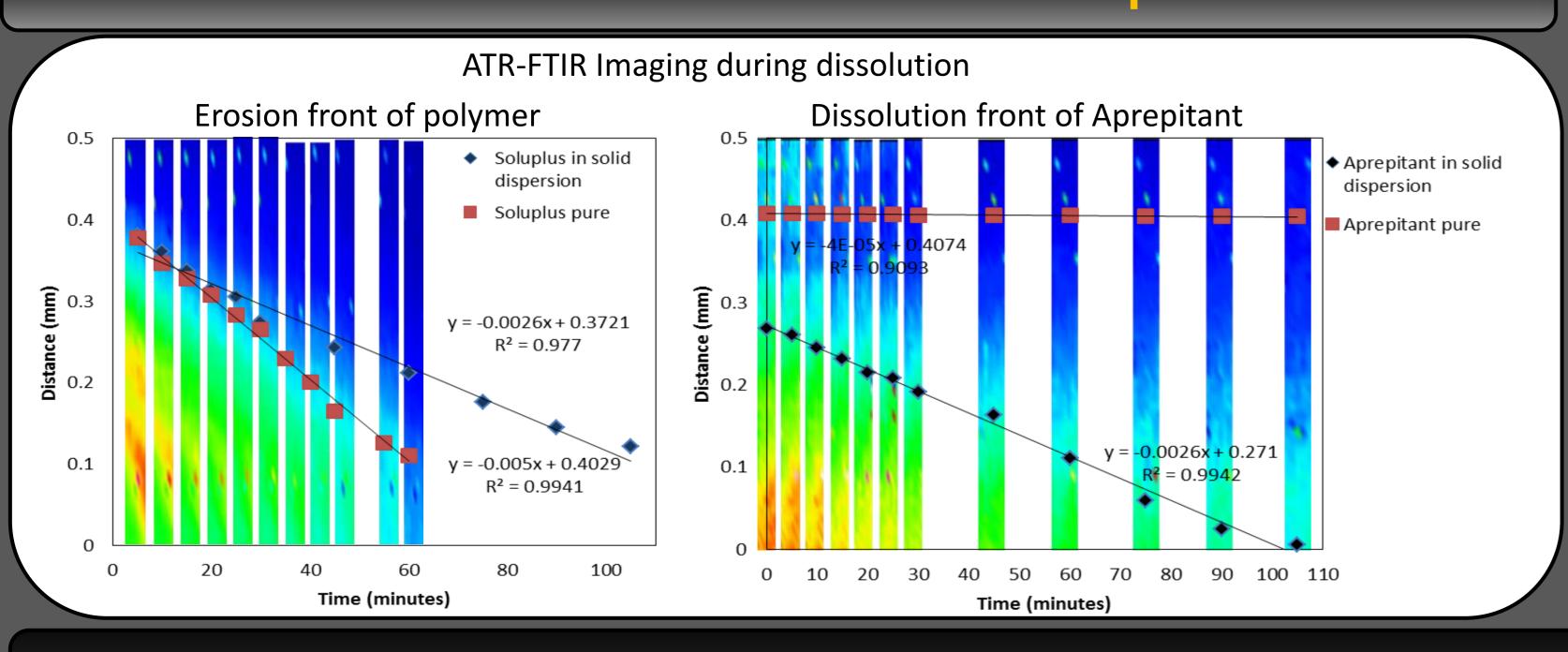
ATR-FTIR imaging recognizes crystallization in solid dispersion system

Absorbance images during dissolution Solid dispersions Precipitation 20 min 45 min 20 min 45 min Soluplus IR spectra of crystalline and amorphous Aprepitant Precipitation of Aprepitant during dissolution Shift of band Presence of peak 1000 cm⁻¹ Shift of peak -1124 cm⁻¹ to 1134 cm⁻¹ 10 min Shift of band 15 min Crystalline API 25 min **Amorphous API** 75 min --- 300 min 900 1200 1100 1000 1150 1000 1200 1100 1050 Wavenumber cm⁻¹ Wavenumber cm⁻¹

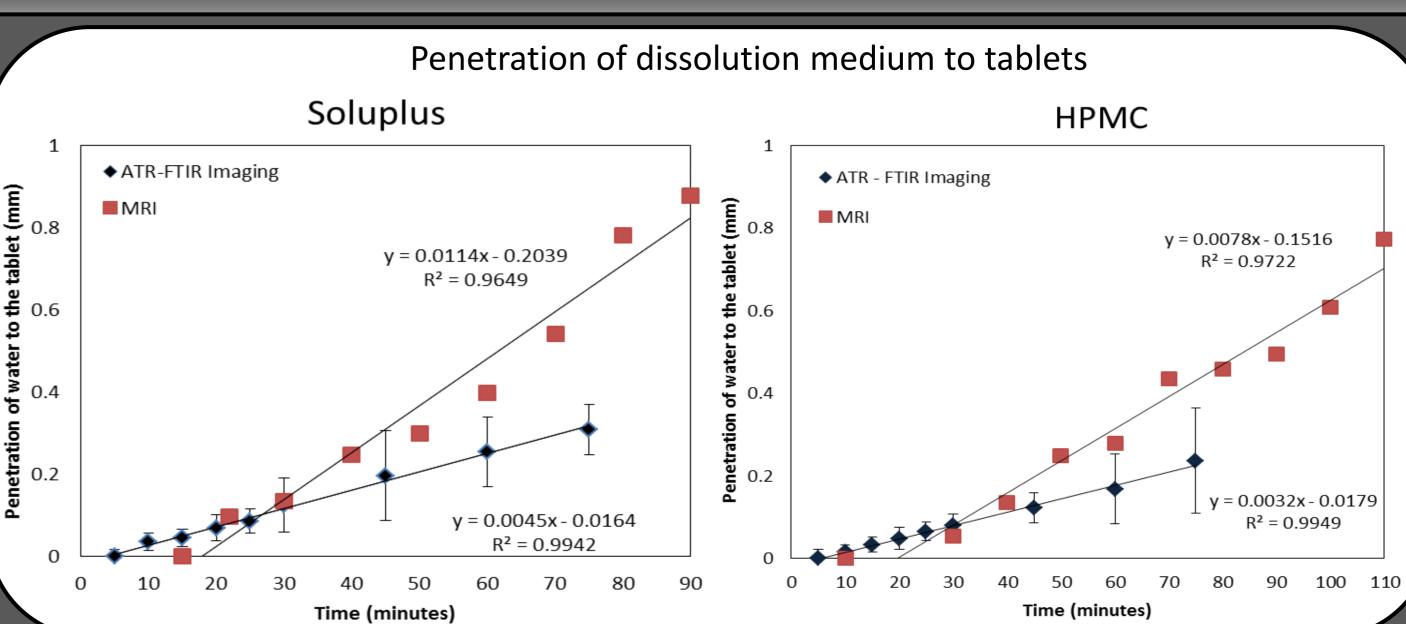
MRI recognizes different mechanism of water penetration to solid dispersion and physical mixture



Different dissolution rate in solid dispersion



Comparison of MRI and ATR-FTIR Imaging



Conclusions

Dissolution process of poorly soluble drug in solid dispersion

- **Soluplus** stabilizes amorphous form by gradual slow dissolution, suitable for dissolution of poorly soluble drugs
- **PVP** does not stabilize amorphous form during dissolution, suitable for fast dissolution
- Precipitation detected by visual observation and IR spectra

Acknowledgment:

Financial support from the Specific University Research (MSMT 2013/2014) is gratefully acknowledged.

Comparison of MRI and ATR-FTIR Imaging

- Slower water penetration in Absorbance Imaging caused by limitation of crystal
- Penetration of water to Soluplus 2.53x faster in MRI
- Penetration of water to HPMC 2.43x faster in MRI
 - ¹ Kazarian S.G., Chan K. L. A., Macromolecules, 2003, 36, 9866-9872.
 - ² Kazarian S.G., Ewing A. V., Expert Opin. Drug Deliv., 2013, 10(9), 1207-1221